

NITROGEN BRIDGEHEAD COMPOUNDS PART 12¹. THE REACTION OF THE ETHYL 6-METHYL-4-OXO-6,7,8,9-TETRAHYDRO-4H-PYRIDO [1,2-a] PYRIMIDINE-3-CARBOXYLATE WITH IMINIUM CHLORIDES.

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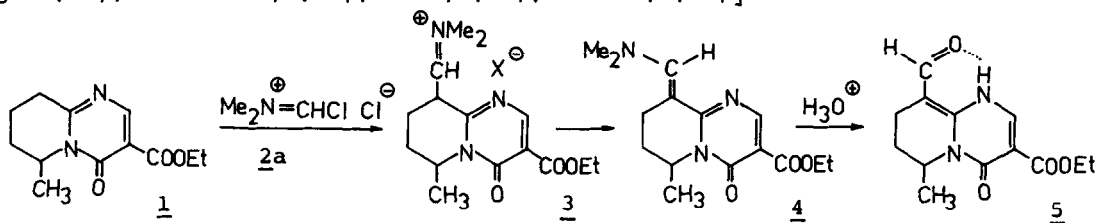
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The 9-formyl- and 9-ester derivatives /5,9/ can easily be prepared from the title pyrido [1,2-a] pyrimidine /1/ with iminium chlorides /2a,b/ via the 9-/dimethyl-iminium-methyl/-derivatives /3,6/.

Pyrido [1,2-a] pyrimidines have increasing pharmacological significance². Earlier we reported³ that the title compound /1/ contains an activated methylene group in the 9-position. This observation offers the possibility of further chemical modification of the parent pyrido-pyrimidines of type 1 introducing different synthones into the 9-position. In this way we aimed to obtain new derivatives, possibly with favourable pharmacological properties.

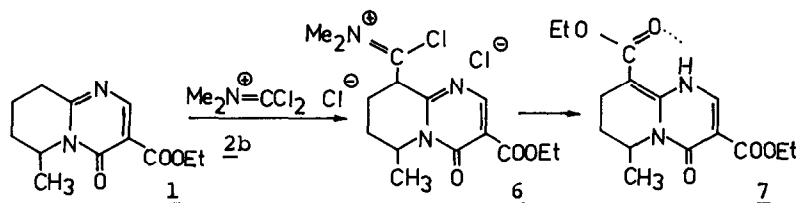
We have investigated the reactivity of the pyrido-pyrimidine /1/ towards iminium-chlorides /2a,b/. 1 with DMF and COCl₂ /molar ratio=1:1:1/ gave 3 in boiling anhydrous methylene chloride. Yield 93 %. /X=Cl mp:211°C decomp./ The 9-amino-methylene derivative /4/ was obtained from 3 with aqueous NaOAc. Yield 80%. 4 was also prepared treating 1 /1 mol/ with POCl₃ /2 mol/ and DMF /10 mol/ at ambient temperature for 2 hrs; yield 78%. [4 mp:136-7°C /EtOH/; uv /EtOH/: λ_{max} 397 /lgε 4,68/, 299 inf. /3,43/, 275 /3,49/, 230nm /4,23/].



According to ¹H-NMR the 9-amino-methylene derivative /4/ exists in CDCl₃-DMSO-d₆ only as the E isomer [δ_{Me₂N-CH=} 8,31ppm, triplett, ⁴J_{=CH,C(8)H₂} = 1Hz]. The 6-methyl group is in quasiaxial position /J_{6e7e} ≈ J_{6e7a} ≈ 3Hz/ because of 1,3-allylic strain^{4,5}.

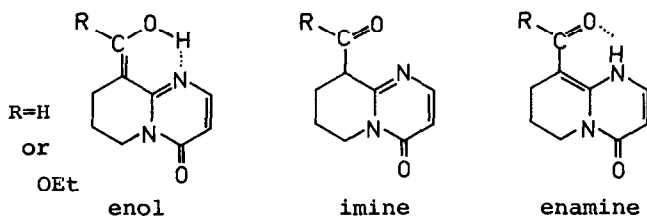
The 9-formyl-derivative /5/ [mp:130-1°C /EtOH/; yield 81%; uv /EtOH/ λ_{max} 358 /lgε 4,34/, 315 inf. /3,86/, 265 /3,49/, 223nm /4,20/] was formed from 3 with water or from 4 with 5% aqueous HCl at 50°C.

1 smoothly reacted with phosgen iminium chloride⁶ /2b/ in boiling anhydrous methylene chloride affording 6 /mp:181-3°C, yield 97%/.



6 treated with EtOH and NaOAc gave the 9-ester /7/ mp:144-6°C, yield 71%; uv /EtOH/ λ_{\max} 335 /lg ϵ 4,11/, 298nm /4,35/.

Compounds 5 and 7 may exist in three tautomeric forms: enol—imine—enamine.



Recently the ethyl-2-methyl-4-oxo-tetrahydro-4H-pyrido [1,2-a] pyrimidine-9-carboxylate^{7,8} and the 2-phenyl derivative⁸ was found to exist in CDCl₃ and in DMSO-d₆ as a 20:80 mixture of the imine-enamine tautomers. With com-

pounds 5 and 7 only the enamine tautomer could be detected in both solvents. The C(2)H proton gave rise to a doublet with $J_{N(1)H-C(2)H}$ =5 and 7Hz, for 5 and 7 respectively. The enamine structure is stabilized by strong intramolecular hydrogen bonding / δ_{NH} =14,1ppm for 5 and 11,52ppm for 7/ . The 6-methyl group is quasi-axial in both compounds.

These synthonies can be useful also for protecting the 9-methylene group of compounds of type 1. The formyl derivative /5/ treated with 5% HCl readily and nearly quantitatively retransforms into the parent 1. The formyl and the ester groups can also be removed by treatment with 5% NaOH, when after acidification the 3-carboxyl derivative of 1 is gained.

¹H-NMR Chemical shifts, δ ppm, /Intensity/ JEOL-PS-100 /in CDCl₃-TMS/

Com- pound	2-H	6-H	7-CH ₂	8-CH ₂	6-Me	1-NH	=CH-	-OCH ₂ -	Me	NMe ₂
<u>4</u>	8,41s /1H/	5,16m /1H/	1,6-2,05m /2H/	2,7-2,9m /2H/	1,28d /3H/	-	8,19t ^a /1H/	4,31q /2H/	1,39t /3H/	3,25s /6H/
<u>5</u>	8,23d ^b /1H/	5,00m /1H/	1,5-2,10m /2H/	2,4-2,85m /2H/	1,25d /3H/	14,5br /1H/	8,92d ^c /1H/	4,23q /2H/	1,42t /3H/	-
<u>7</u>	8,00d ^d /1H/	5,10m /1H/	1,55-2,1m /2H/	2,45-2,8m /2H/	1,23d /3H/	11,52d /1H/	-	4,26q 4,31q /2x2H/	1,36t 1,40t /2x3H/	-

a/ $^4J_{CH-C(8)H_2} \approx 1\text{Hz}$; b/ $^3J_{C(2)H-N(1)H} \approx 5\text{Hz}$; c/ $^5J_{CH-N(1)H} \approx 1\text{Hz}$; d/ $^3J_{C(2)H-N(1)H} \approx 7\text{Hz}$

REFERENCES: 1./ Part 11: I. Hermezc et al.:Tetrahedron Lett. in press; 2a./ J. Knoll, S. Fürst and Z. Mészáros: Arzneim.Forsch. /Drug Res./ 1971,21,719; b./ J. Knoll, Z. Mészáros and G. Zsila: ibid. in press; c./ I. Hermezc et al.:ibid. in press; d./ P.F. Juby: 4.122.274 USA patent; 3a./ Z. Mészáros et al.:Arzneim. Forsch. /Drug Res./ 1972,22,815; b./ G. Náray-Szabó, I. Hermezc and Z. Mészáros: J.C.S. Perkin I 1974,1753; 4./ F. Johnson: Chem.Rev. 1968,68,375; 5./ G. Tóth, I. Hermezc and Z. Mészáros: in press; 6./ H.G.Viehe and Z. Janousek: Angew.Chem. 1971,83,614; 7./ H.L. Yale and E.R. Spitzmiller: J. Heterocyclic Chem. 1976,13, 797; 8./ H. Wamhoff and L. Lichtenthäler: Chem.Ber. 1978,111,2813.